

Absolute gas thermometry using IR emission spectroscopy

Ared Cezairliyan Best Paper Award of the Int. Journal of Thermophysics

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What is IR emission spectroscopy?



Radiative Transfer in 1D uniform T case:

 $I_{\eta}(v) = L(v) \ \varepsilon(v) \ or$

- $I_{\eta}(v) = L(v) \ (1 \exp(-k(v)L))$
- \Box *L*(*v*) is Planck's function
- \Box k(v) is an (molecular) absorption coefficient
- □ *L* dimention
- \Box $\varepsilon(v)$ emissivity, if $\varepsilon(v) = 1$ then BB

Detector:

- 1D: (handheld) pyrometer or FTIR Spectrometer
- 2D: Imaging Camera (array) or Hyperspectral Imaging System

L: from few cm to few m "Gas" ↔ "Detector" distance: close or far-away

Two "magic" wavelengths: 4.3 μm (2326 cm $^{-1}) and 3.9 <math display="inline">\mu m$ (2564 cm $^{-1})$

If $1 \ll k(v)L$ then $I_{\eta}(v) \approx L(v)$ and

> CO_2 band at 4.3 µm can be used for T_{gas} measurements

No "if" then (inverse) RHT equation should be solved with use spectra modeling tools (HITEM/HITRAN databases)

3.9 μ m can be used for particles and/or surface temperature measurements (assuming some known $\varepsilon(v)$)

Requires a calibration with a reference (calibrated) BB source

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Why IR emission spectroscopy:

- ✓ non-intrusive
- ✓ fast

(Very) high price-range

- ✓ can be used for *in situ* process control
- ✓ can be realized in a *mid-price* range

Where:

Globally (broad spectral range):

Gas and particle temperatures

Locally (narrow spectral range):

- □ Band-shapes P/T dependent
- Gas temperature





(Very) low price-range



BUT:

Various optical emission spectroscopy methods need a validation with a known temperature and composition source traceable to ITS-90 International Journal of Thermophysics (2019) 40:99 https://doi.org/10.1007/s10765-019-2557-6



Validation of Emission Spectroscopy Gas Temperature Measurements Using a Standard Flame Traceable to the International Temperature Scale of 1990 (ITS-90)

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The aims:

- ✓ to develop a portable flame temperature standard, calibrated via the Rayleigh scattering thermometry technique, traceable to ITS-90, with an uncertainty of 0.5 % of temperature (k = 1).
- $\checkmark\,$ to use standard flame for validation of
 - hyperspectral imaging FTIR spectrometer (2D species and temperature maps)
 - high-precision single line-of-sight FTIR spectrometer



The NPL portable standard flame

"Region of hot gas of known temperature and species"



The NPL standard flame:

- Hencken Burner diffusion flamelets
- **Propane** / air flame $\{0.8 < \phi < 1.4\}$
- Low uncertainty flowmeters $U_r(flow) < 0.5\%$
- Known post flame composition
- Traceable to ITS-90
- Portable!
- Reproducible temperature $U_r(\phi) = 0.5\%$

$$\phi = \frac{\left(V_{fuel}/V_{air}\right)}{\left(V_{fuel}/V_{air}\right)_{Stoichiometric}}$$

Stoichiometric → fuel/air ratio for a balanced reaction (i.e. no excess oxygen)









Burner performance

Temperature profile example, $\phi = 1.0$ (Rayleigh scattering: point measurements)





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- Flame is flat over +/- 10 mm region
- Knowledge of the profile useful for line of sight techniques



HAB = Height Above Burner



Temperature uncertainty

Source	Type	Distr.	Size / ± %	Multiplier	Sensitivity Coefficient	Size (10) /%
Molar refractivity data	В	Rect	0.20	0.58	1.00	0.12
Flow-meter uncertainty	B	Rect	1.00	0.58	0.40	0.23
Chemical equilibrium assumption	B	Rect	0.30	0.58	1.00	0.17
Air calibration PRT	A	Norm	0.05	1.00	1.00	0.05
Background scattered signal	A	Norm	0.10	1.00	0.50	0.05
Laser stability	A	Norm	0.20	1.00	1.00	0.20
Inlet air temperature (15-25 °C)	В	Rect	3.00	0.58	0.10	0.17
Atmospheric pressure	B	Rect	5.00	0.58	0.05	0.15
Gas purity	В	Rect	2.50	0.58	0.05	0.07
Flame temperature reproducibility	A	Norm	0.20	1.00	1.00	0.20
Total uncertainty (combined in quadrature)						0.5 %

Table A1 Uncertainty budget for the temperature 20 mm above the centre of the NPL STD flame.

• Long term stability = 0.2 % of T

Combined uncertainty (k=1) = 0.5 % of T





European project to enhance process efficiency through improved temperature measurement:

- STD flame calibrated at NPL
- Partner organisations developing novel optical combustion thermometers

- STD flame circulated to partners
- Comparison of techniques publication
- NPL facility available for others



EMPRESS - Enhancing Process Efficiency through Improved Temperature Measurement: http://journals.sagepub.com/doi/pdf/10.1177/0020294016656892

Hyper-spectral IR imaging at UC3M



FTIR Hyperspectral Imaging System

- \checkmark Operates in the MIR (2 μm to 5.5 $\mu m)$
- ✓ Michelson interferometer, coupled to an InSb focal plane array (320 x 256 pixels)
- ✓ Similar to FTIR spectrometer but provides 2D maps of species and temperature
- ✓ Measures the emitted power from the CO₂, CO and H₂O bands (post flame region)
- ✓ Comparison with HITEMP-2010 synthesised spectra → determine the flame temperature









Hyper-spectral IR imaging: data analysis and modelling



Example measurement and spectral fit

- Model accounts for absorption in the ambient air
- Temperature profile assumed to be 'top-hat' (out of the page)
- Good agreement between measured and synthesised spectra







Measurements on the NPL STD flame

- $\phi = 1.0$, propane/air flame
- Temperature, CO₂ and CO are measured





2000

-20,0

-15.0

-10,0



0,0 Rx (mm)

-5,0

10,0

5,0

15,0

20,0

Precision IR emission spectroscopy







Measurement principle

• The flame spectral intensity at wavenumber η measured at L is given by:

$$I(\eta) = \int_0^L \kappa_\eta I_{B\eta}(T) e^{-\int_s^L \kappa_\eta ds'} ds$$

- Where:
 - $I_{B\eta}(T)$ is the Planck function
 - κ_{η} is the spectral absorption coefficient

Determining the temperature profile

- Measure the emission spectra
- Estimate temperature profile $T_{est}(s)$ and calculate κ_{η} from HITEMP2010
- Synthesise the emission spectra
- Minimise the differences between the two
- Determine the true temperature profile $T_{true}(s)$





Assumed to be a smooth

Symmetric or asymmetric function



Validation of HITEMP spectral database



Modelled spectra agree very well with measured

Differences between measured/modelled:

- \square H₂O/CO (1500-2000cm⁻¹) = 0.63%
- \Box CO₂/CO (2160-2400cm⁻¹) = -0.17%
- \Box CO₂ band is most sensitive to T(x) profile
- \square H₂O band is less sensitive

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Comparison with Hyper-Spectral measurements



> An overall excellent agreement

 \succ No CO₂ self-absorption (ambient air)



Verification of NPL temperature profiles

- > NPL temperature profile is input parameter for modelled HITEMP2010 spectra
- NPL profile has been improved
- > Example 2: $\phi = 0.8$, HAB = 20 mm



Inspired by industry: NOx reduction in SNCR systems





From Bernd von der Heide (2008), Mehldau & Steinfath Umwelttechnik GmbH

NOx <u>Selective Non-Catalytic Reduction (SNCR)</u> on power plants and waste incinerators:

- > Narrow temperature range (nature of the process): $NH_3 + NO \rightarrow N_2 + H_2O$
- Ammonia/urea consumption optimization (costs)
- NOx high-efficiency removal (environment, pollution)
- Accurate in situ T_{gas} measurements is a must
- 2D temperature profiles with a sweeping technique (several line-of-sights measurements combined in 2D plot)
- Fast T-profiles retrievals
- Possibility for a "moderate" system cost
- > Only 1x access point





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Machine learning approach for retrievals of temperature profiles









Predicted mean temperatures within 10 mm from the center of the burner differ by

▶ φ=1: 0.23%

- **□** *φ*=1.4: 1.88%

from the measured temperatures.



Target: an overall temperature uncertainty below 0.5 %



Waste insinirator: FTIR 8cm⁻¹ fast scanning

In Situ measurements at a waste incinerator

Challenges:

- ✓ Keep optics clean (CO₂ self-absorption)
- ✓ HITEMP2010 "weak" line intensities those become "strong" at long L (10 m)

- ✓ Reference CO₂/H₂O high-T spectra can be in house measured
- ✓ There are (natural) temporal T_{gas} -variations due to turbulence (from ± 20 °C to ± 60 °C)

> 1% from 1000°C is 10°C ($\Delta T = 50$ °C window for SNCR)



Conclusions

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